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# Intense blue photoluminescence of the Tm<sup>3+</sup>/Yb<sup>3+</sup> co-doped single-crystalline hexagonal phase NaYF<sub>4</sub> nanorods

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#### ABSTRACT

In this paper,  $Tm^{3+}/Yb^{3+}$  co-doped NaYF<sub>4</sub> single-crystalline nanorods were prepared via a rational hydrothermal method using oleic acid as a stabilizing agent. The microstructure analysis revealed that the as-prepared samples presented a pure hexagonal phase NaYF<sub>4</sub> structure with high quality rod-like morphology by means of X-ray diffraction (XRD) and transmission electron microscopy (TEM). Under 980 nm excitation, the intense blue and weaker red upconversion emissions of the NaYF<sub>4</sub>:  $1^{8}Tm^{3+}/x^{8}Yb^{3+}$  (x=2, 5, 10 and 15) samples centered at 450, 477, 649 and 690 nm were observed, which were attributed to the  $^{1}D_{2} \rightarrow ^{3}F_{4}$ ,  $^{1}G_{4} \rightarrow ^{3}H_{6}$ ,  $^{1}G_{4} \rightarrow ^{3}H_{4}$  and  $^{3}F_{3} \rightarrow ^{3}H_{6}$  transitions of  $Tm^{3+}$ , respectively. Moreover, the photograph of the  $1^{8}Tm^{3+}/10^{8}Yb^{3+}$  co-doped NaYF<sub>4</sub> nanorods exhibited a strong eye-visible blue emission. The upconversion luminescence mechanisms for the  $Tm^{3+}/Yb^{3+}$  co-doped hexagonal phase NaYF<sub>4</sub> nanorods were analyzed. In addition, the effect of  $Yb^{3+}$  ions concentrations on the photoluminescence properties of the  $Tm^{3+}/Yb^{3+}$  co-doped hexagonal phase NaYF<sub>4</sub> nano-rods were investigated in detail.

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# 1. Introduction

In recent years, the rare earth (RE) doped optical materials have been extensively investigated due to their potential applications in many fields, such as color display, optical data storage, sensor, laser and optical amplifier for communication [1–5]. Upconversion is the generation of higher energy light such as ultraviolet or visible light from lower energy radiation, usually near-infrared or infrared via two photon or multi-photon mechanism [6]. It is well known that, as the host for the luminescent RE, fluoride is preferable over oxide mainly for the lower phonon energy to avoid non-radiative transition of RE ions [7]. To the best of our knowledge, among the reported fluoride upconversion materials, hexagonal phase NaYF4 is one of the most efficient upconversion host material for visible upconversion luminescence [8]. In recent years, one dimensional nanoscale materials, such as nanowires, nanotubes, nanobelts and nanorods, have attracted much interest due to their potential applications in fabricating nanoscale optoelectronics [9,10], electronics [11,12], lasers [9,13] and biological labels [14], etc. Many unique and fascinating properties have already been demonstrated for this kind of materials, such as high luminescence efficiency, enhancement of thermoelectric figure of merit and a lowered lasing threshold [15,16]. Hence, it is important to synthesize highly uniform and monodispersed hexagonal phase NaYF<sub>4</sub> nanocrystals with one dimensional rod-shaped structure, as mentioned above. However, most of the previous reports were mainly focused on the preparation of NaYF4 nano- and micro-crystals morphologies [17–22]. Only limited reports were devoted to synthesizing hexagonal phase NaYF4 nanocrystals with rod-like shape. In this paper,  $Tm^{3+}/Yb^{3+}$  co-doped hexagonal phase NaYF4 single-crystalline nanorods with high quality were prepared via a rational hydrothermal method using oleic acid as a stabilizing agent. Intense blue emission can be achieved from this  $Tm^{3+}/Yb^{3+}$  co-doped hexagonal phase NaYF4 nanorods. And the effect of  $Yb^{3+}$  concentrations on upconversion property of the  $1\%Tm^{3+}/x\%Yb^{3+}$  (x=2,5,10 and 15) co-doped hexagonal phase NaYF4 single-crystalline nanorods was studied in detail.

# 2. Experimental

The Tm3+/Yb3+ co-doped NaYF4 nanorods were prepared with the following compositions (in mol %): NaYF<sub>4</sub>:  $1\%\text{Tm}^{3+}/x\%\text{Yb}^{3+}$  (x = 2, 5, 10 and 15). All the rareearth oxides were 99.99% purity and other raw materials are of analytical reagent grade. Rare earth nitrate RE(NO<sub>3</sub>)<sub>3</sub> (RE=Y, Yb and Tm) solutions with 0.5 mol L<sup>-1</sup>, 0.5 mol L<sup>-1</sup> and 0.1 mol L<sup>-1</sup> were prepared by dissolving the corresponding rare earth oxide (99.99%) in nitric acid at elevated temperature respectively and excess nitric acid was removed by evaporation. A typical synthesis of  $Tm^{3+}/Yb^{3+}$  co-doped NaYF<sub>4</sub> nanorods were performed by a rational hydrothermal method using oleic acid as a stabilizing agent [23,24-26]. For a typical synthesis, 1.2 g NaOH, 2 mL distilled water, 9 mL ethanol, and 20 mL oleic acid were mixed together under vigorously stirring to form a homogeneous solution. Next, 1 mmol (total amounts) of RE(NO<sub>3</sub>)<sub>3</sub> (RE = Y, Yb and Tm) and 8 mL NaF (1.0 M) aqueous solution were added to the solution under vigorously stirring. The resulting mixture was stirred for another 30 min. The resulting solution was then transferred into a 50 mL stainless Teflon-lined autoclave, sealed, and heated at 190 °C for 24 h. The system was then allowed to cool down to room temperature naturally and the products were deposited at the bottom of the vessel. The final products can be well-dispersed

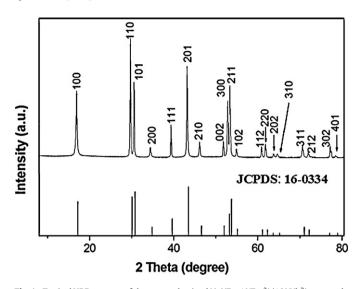
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in a nonpolar solvent such as cyclohexane and aggregated by adding polar solvent such as ethanol. The obtained nanorods were washed with ethanol and water to remove oleic acid and other remnants, and then dried in air at  $60\,^{\circ}\text{C}$  for  $6\,\text{h}$ 

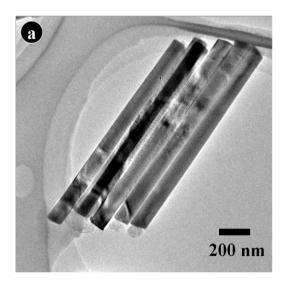
The crystal phase of the as-obtained samples was recorded by a D/max- $\gamma$ A System X-ray diffractometer at 40 kV and 40 mA with Cu-K $\alpha$  radiation ( $\lambda$  = 0.15418 nm). The scan was performed in the  $2\theta$  range from  $5^{\circ}$ –80° with a scanning rate of 0.04°/s and step size of 0.04°. The morphology and microstructure of the as-prepared samples were further characterized by TEM and the high resolution transmission electron microscopy (HRTEM) assays using a JEM-2100 microscope operated at 200 kV. The upconversion emission spectra at room temperature were recorded by a spectrophotometer (R500) under the excitation of 980 nm laser diode (LD). All the above measurements were carried out at room temperature.

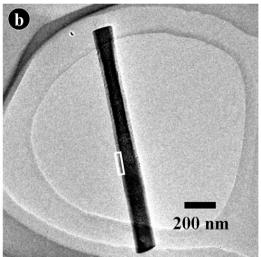
# 3. Results and discussion

NaYF $_4$  exists in two polymorphs at ambient pressure: cubic phase and hexagonal phase. The phase compositions of the asprepared products were detected by the powder XRD pattern. The XRD pattern of NaYF $_4$ : 1% Tm $^{3+}$ , 10% Yb $^{3+}$  samples obtained from the hydrothermal condition at 190 °C for 24 h is shown in Fig. 1. The peak positions and intensities of the XRD pattern matched well with the reference hexagonal phase NaYF $_4$  (JCPDS 16-0334). No other impurity peaks were detected. In addition, the well-defined diffraction peaks demonstrated that the as-prepared samples had high crystalline nature. Similar results were observed for the 1%



**Fig. 1.** Typical XRD pattern of the as-synthesized NaYF<sub>4</sub>:  $1\%\text{Tm}^{3+}/10\%\text{Yb}^{3+}$  nanorods. Tm<sup>3+</sup>/ $x\%\text{Yb}^{3+}$  (x = 2, 5 and 15) doped samples. However, the preferred growth direction of the nanorods is not directly reflected in the XRD patterns, which may be related to the relative small aspect ratio of the nanorods.





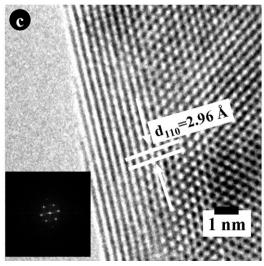
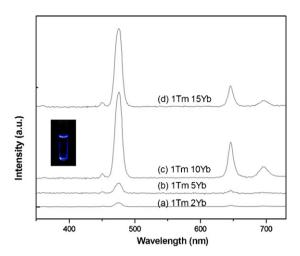


Fig. 2. The typical TEM and HRTEM images of the NaYF<sub>4</sub>:  $1\%\text{Tm}^{3+}/10\%\text{Yb}^{3+}$  nanorods: (a) low magnification image, (b) the individual nanorod, (c) the corresponding HRTEM image taken from the area marked by a white rectangle. The inset of (c) is the corresponding FFT image.



**Fig. 3.** Dependence of upconversion emission spectra on  $Yb^{3+}$  concentration of NaYF<sub>4</sub>:  $1\%Tm^{3+}/x\%Yb^{3+}$  samples: (a)  $1 mol\%Tm^{3+}$ ,  $2 mol\%Yb^{3+}$  ions; (b)  $1 mol\%Tm^{3+}$ ,  $5 mol\%Yb^{3+}$ ; (c)  $1 mol\%Tm^{3+}$ ,  $10 mol\%Yb^{3+}$ ; (d)  $1 mol\%Tm^{3+}$ ,  $15 mol\%Yb^{3+}$ . The inset shows the photograph of 1 wt% colloidal solution of as-prepared NaYF<sub>4</sub>:  $1\%Tm^{3+}/10\%Yb^{3+}$  samples dispersed in cyclohexane under the excitation of 980 m excitation laser diode with power density of  $3 W/cm^2$ .

The morphology and microstructure of the samples were characterized by TEM and HRTEM. Fig. 2a showed the typical TEM image of the as-prepared NaYF<sub>4</sub>: 1% Tm<sup>3+</sup>/10%Yb<sup>3+</sup> samples and Fig. 2b is the typical TEM image of an individual NaYF<sub>4</sub> nanorod. From these TEM observations, it can be seen that the sample exhibits rodlike morphology with smooth wall. Furthermore, these nanorods display uniform morphology and high quality. The length of the as-prepared nanorods is about 1500 nm and the diameter is about 90 nm, as estimated from the TEM images, and the aspect ratio is about 16. Fig. 2c shows the typical HRTEM image taken from the white rectangle area of the individual nanorod. As demonstrated in Fig. 2c, the as-prepared NaYF<sub>4</sub> nanorods exhibited highly crystalline nature, which is structural uniformity and free of the defects. The measured interplanar distance is about 2.96 Å, corresponding to the (110) lattice plane of the hexagonal phase NaYF<sub>4</sub>, which is consistent with the XRD result. The inset of Fig. 2c is the corresponding fast fourier transform (FFT) image. The regular diffraction spots of FFT image recorded along the [001] zone axis unambiguously demonstrate the single nanorod has single-crystalline nature. The diffraction pattern can be readily indexed as the hexagonal phase NaYF<sub>4</sub>, which is also well consistent with the XRD analysis. Based on the HRTEM analysis, the preferred growth direction of the as-prepared nanorods is along the [110] direction.

The upconversion photoluminescence of different compositions of Tm<sup>3+</sup>/Yb<sup>3+</sup> co-doped hexagonal phase NaYF<sub>4</sub> nanorods are shown in Fig. 3. Under the excitation of a 980 nm LD, the blue and red emissions centered at 450, 477, 649 and 690 nm were observed, which were attributed to the  $^1D_2 \rightarrow {}^3F_4$ ,  $^1G_4 \rightarrow {}^3H_6$ ,  $^1G_4 \rightarrow {}^3F_4$  and  $^3F_3 \rightarrow {}^3H_6$  transitions of Tm $^{3+}$ , respectively. The inset of Fig. 3c shows the photograph of 1 wt% colloidal solution of as-prepared NaYF<sub>4</sub>: 1%Tm<sup>3+</sup>/10%Yb<sup>3+</sup> nanorods dispersed in cyclohexane under the excitation of 980 nm LD with power density of 3 W/cm<sup>2</sup>. As demonstrated in the photography image, the bright blue upconversion luminescence can be observed by naked eyes. It is also noted that the intensities of these bands increase with increasing the concentration of Yb3+ and further decrease when the concentration of Yb<sup>3+</sup> ions reaches at 15 mol%, which may be attributed to the concentration quenching [27]. According to the simplified energy level diagram shown in Fig. 4, the possible upconversion mechanisms have been discussed in detail. Excitation photons ( $\lambda = 980 \text{ nm}$ ) are strongly absorbed by isolated Yb3+ ions because Yb3+ ions have larger absorption cross section and Tm<sup>3+</sup> ions have no correspond-

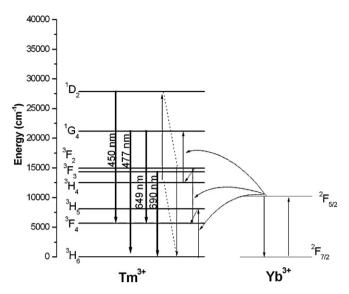


Fig. 4. Simplified energy-level diagram of Tm<sup>3+</sup> and Yb<sup>3+</sup>.

ing energy level. These ions act as sensitizers and this absorption process is strongly dependent on the distance between neighboring ions. After absorption of photons by Yb<sup>3+</sup>, energy transfer (ET) takes place between Yb<sup>3+</sup> and Tm<sup>3+</sup>. Under 980 nm excitation, the Yb<sup>3+</sup> ions are excited from the  $^2F_{7/2}$  level to the  $^2F_{5/2}$  level, then transfer their energies to the nearby  $Tm^{3+}$  ions, then three successive ET from Yb<sup>3+</sup> to Tm<sup>3+</sup> populate the  $^3H_5$ ,  $^3F_2$ , and  $^1G_4$  levels of Tm<sup>3+</sup> [28]. Consequently, the blue emission centered at 477 nm is generated by the  ${}^{1}G_{4} \rightarrow {}^{3}H_{6}$  transition of Tm<sup>3+</sup> ions. While, the red emissions at 649 and 690 nm are due to the  ${}^1G_4 \rightarrow {}^3F_4$  and  ${}^3F_3 \rightarrow {}^3H_6$ transitions of  $Tm^{3+}$  ions, respectively. Due to the large energy mismatch between  $^1G_4$  and  $^1D_2$  levels of  $Tm^{3+}$  (about 3500 cm $^{-1}$ ), the  $^1\text{G}_4 \rightarrow ^1\text{D}_2$  transition of  $\text{Tm}^{3+}$  cannot be directly achieved by the absorption of the fourth photon from Yb³+ via ET process. Therefore, the cross relaxation (CR) process of  ${}^3F_2 + {}^3H_4 \rightarrow {}^3H_6 + {}^1D_2$  between Tm<sup>3+</sup> ions may alternatively play an important role in populating <sup>1</sup>D<sub>2</sub> level [29,30], which subsequently resulting in the blue emission band centered at 450 nm. However, compared with the blue emission band centered at 477 nm, the intense of the blue emission centered at 450 nm is much weaker, which is ascribed to the poor CR process of Tm<sup>3+</sup> ions.

# 4. Conclusions

A series of  $1\%\text{Tm}^{3+}/x\%\text{Yb}^{3+}$  (x = 2, 5, 10, 15) co-doped hexagonal phase NaYF<sub>4</sub> single-crystal nanorods with high quality were fabricated by a facile hydrothermal method. The intense blue emission centered at 477 nm of Tm<sup>3+</sup>/Yb<sup>3+</sup> co-doped hexagonal phase NaYF<sub>4</sub> nanorods were observed under the excitation of a 980 nm laser diode. The influence of Yb3+ concentration on all the emissions is obvious, which indicates the Yb3+ ion plays an important role in the energy transfer process. With increasing the Yb<sup>3+</sup> contents, the upconversion emission intensity gradually increased. However, it decreased when the doping of Yb<sup>3+</sup> reached to 15 mol%, owing to the effect of concentration quenching. More importantly, the intense eye-visible blue luminescence can be observed from the 1%Tm<sup>3+</sup>/10%Yb<sup>3+</sup> co-doped NaYF<sub>4</sub> nanorods under the excitation of a 980 nm laser diode with power density of 3 W/cm<sup>2</sup>. It is therefore expected that these single-crystal and high quality onedimensional nanorods with intense eye-visible blue upconversion luminescence may have potential applications in visible solid-state lasers and biolabels.

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